1. If a beer starts at an ethanol concentration of 0.5 M and you prepare a 1:200 dilution of the beer with water, what is the concentration of ethanol in the dilution?

| Tube | Buffer <br> (without <br> semicarbazide) | Buffer <br> (with $75 \mathbf{~ m M ~}$ <br> semicarbazide) | $\mathbf{1 6 ~ m M}$ <br> $\mathbf{N A D}^{+}$ | $\mathbf{1 2 0 0}$ <br> units/mL <br> ADH | EtOH test <br> solution | Unknown <br> solution | $\mathbf{d H}_{\mathbf{2}} \mathbf{O}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1}$ | $2,625 \mathrm{uL}$ | - | 150 uL | 75 uL | - | - | 150 uL |
| $\mathbf{2}$ |  | $2,625 \mathrm{uL}$ | 150 uL | 75 uL | - | - | 150 uL |
| $\mathbf{3}$ | $2,625 \mathrm{uL}$ | - | 150 uL | 75 uL | 150 uL | - | - |
| $\mathbf{4}$ | - | $2,625 \mathrm{uL}$ | 150 uL | 75 uL | 150 uL | - | - |
| $\mathbf{5}$ | $2,625 \mathrm{uL}$ | - | 150 uL | 75 uL | - | 150 uL | - |
| $\mathbf{6}$ | - | $2,625 \mathrm{uL}$ | 150 uL | 75 uL | - | 150 uL | - |

2. Assume that the "unknown solution" of ethanol is the 1:200 dilution from question \#1 and that the "EtOH test solution" is at 3 mM . Rank the expected absorbance at 340 nm from highest absorbance to lowest absorbance for tubes 3 through 6.
